

# 2014 Consumer Confidence Report

Water System Name: Chemeketa Park Mutual Water Company Report Date: June 26, 2015

*We test the drinking water quality for many constituents as required by state and federal regulations. This report shows the results of our monitoring for the period of January 1 - December 31, 2014 and may include earlier monitoring data.*

**Este informe contiene información muy importante sobre su agua potable. Tradúzcalo ó hable con alguien que lo entienda bien.**

Type of water source(s) in use: Mountain Creek Surface Water

Name & general location of source(s): Moody Gulch Creek intake is located at the north entrance of Ogallala Ogallala Warpath at Old Cruz Hwy. The Los Gatos Creek Pump station is directly below Chemeketa Park at the end of Assiniboine Trail.

Drinking Water Source Assessment information: \_\_\_\_\_

Time and place of regularly scheduled board meetings for public participation: Board meetings are held the 2nd Thursday of each month at 8:00 pm in the Chemeketa Park Clubhouse.

For more information, contact: Tyler Boswell, (Certified System Operator II) Phone: (408) 590-9715

## TERMS USED IN THIS REPORT

**Maximum Contaminant Level (MCL):** The highest level of a contaminant that is allowed in drinking water. Primary MCLs are set as close to the PHGs (or MCLGs) as is economically and technologically feasible. Secondary MCLs are set to protect the odor, taste, and appearance of drinking water.

**Maximum Contaminant Level Goal (MCLG):** The level of a contaminant in drinking water below which there is no known or expected risk to health. MCLGs are set by the U.S. Environmental Protection Agency (USEPA).

**Public Health Goal (PHG):** The level of a contaminant in drinking water below which there is no known or expected risk to health. PHGs are set by the California Environmental Protection Agency.

**Maximum Residual Disinfectant Level (MRDL):** The highest level of a disinfectant allowed in drinking water. There is convincing evidence that addition of a disinfectant is necessary for control of microbial contaminants.

**Maximum Residual Disinfectant Level Goal (MRDLG):** The level of a drinking water disinfectant below which there is no known or expected risk to health. MRDLGs do not reflect the benefits of the use of disinfectants to control microbial contaminants.

**Primary Drinking Water Standards (PDWS):** MCLs and MRDLs for contaminants that affect health along with their monitoring and reporting requirements, and water treatment requirements.

**Secondary Drinking Water Standards (SDWS):** MCLs for contaminants that affect taste, odor, or appearance of the drinking water. Contaminants with SDWSs do not affect the health at the MCL levels.

**Treatment Technique (TT):** A required process intended to reduce the level of a contaminant in drinking water.

**Regulatory Action Level (AL):** The concentration of a contaminant which, if exceeded, triggers treatment or other requirements that a water system must follow.

**Variations and Exemptions:** Department permission to exceed an MCL or not comply with a treatment technique under certain conditions.

**ND:** not detectable at testing limit

**ppm:** parts per million or milligrams per liter (mg/L)

**ppb:** parts per billion or micrograms per liter (µg/L)

**ppt:** parts per trillion or nanograms per liter (ng/L)

**ppq:** parts per quadrillion or picogram per liter (pg/L)

**pCi/L:** picocuries per liter (a measure of radiation)

**The sources of drinking water** (both tap water and bottled water) include rivers, lakes, streams, ponds, reservoirs, springs, and wells. As water travels over the surface of the land or through the ground, it dissolves naturally occurring minerals and, in some cases, radioactive material, and can pick up substances resulting from the presence of animals or from human activity.

**Contaminants that may be present in source water include:**

- *Microbial contaminants*, such as viruses and bacteria that may come from sewage treatment plants, septic systems, agricultural livestock operations, and wildlife.
- *Inorganic contaminants*, such as salts and metals, that can be naturally-occurring or result from urban stormwater runoff, industrial or domestic wastewater discharges, oil and gas production, mining, or farming.
- *Pesticides and herbicides* that may come from a variety of sources such as agriculture, urban stormwater runoff, and residential uses.
- *Organic chemical contaminants*, including synthetic and volatile organic chemicals that are by-products of industrial processes and petroleum production, and can also come from gas stations, urban stormwater runoff, agricultural application, and septic systems.
- *Radioactive contaminants* that can be naturally occurring or be the result of oil and gas production and mining activities.

**In order to ensure that tap water is safe to drink**, the USEPA and the California Department of Public Health (Department) prescribe regulations that limit the amount of certain contaminants in water provided by public water systems. Department regulations also establish limits for contaminants in bottled water that provide the same protection for public health.

**Tables 1, 2, 3, 4, 5, 7, and 8 list all of the drinking water contaminants that were detected during the most recent sampling for the constituent.** The presence of these contaminants in the water does not necessarily indicate that the water poses a health risk. The Department allows us to monitor for certain contaminants less than once per year because the concentrations of these contaminants do not change frequently. Some of the data, though representative of the water quality, are more than one year old.

**TABLE 1 – SAMPLING RESULTS SHOWING THE DETECTION OF COLIFORM BACTERIA**

Microbiological Contaminants (complete if bacteria detected)	Highest No. of Detections	No. of months in violation	MCL	MCLG	Typical Source of Bacteria
Total Coliform Bacteria	(In a mo.)		More than 1 sample in a month with a detection	0	Naturally present in the environment
Fecal Coliform or <i>E. coli</i>	(In the year)		A routine sample and a repeat sample detect total coliform and either sample also detects fecal coliform or <i>E. coli</i>	0	Human and animal fecal waste

**TABLE 2 – SAMPLING RESULTS SHOWING THE DETECTION OF LEAD AND COPPER**

Lead and Copper (complete if lead or copper detected in the last sample set)	Sample Date	No. of samples collected	90 <sup>th</sup> percentile level detected	No. sites exceeding AL	AL	PHG	Typical Source of Contaminant
Lead (ppb)	12/16/'13	5	0	0	15	0.2	Internal corrosion of household water plumbing systems; discharges from industrial manufacturers; erosion of natural deposits
Copper (ppm)	12/16/'13	5	.057	0	1.3	0.3	Internal corrosion of household plumbing systems; erosion of natural deposits; leaching from wood preservatives

**TABLE 3 – SAMPLING RESULTS FOR SODIUM AND HARDNESS**

Chemical or Constituent (and reporting units)	Sample Date	Level Detected	Range of Detections	MCL	PHG (MCLG)	Typical Source of Contaminant
Sodium (ppm)		27		none	none	Salt present in the water and is generally naturally occurring
Hardness (ppm)		220		none	none	Sum of polyvalent cations present in the water, generally magnesium and calcium, and are usually naturally occurring

\*Any violation of an MCL or AL is asterisked. Additional information regarding the violation is provided later in this report.

**TABLE 4 – DETECTION OF CONTAMINANTS WITH A PRIMARY DRINKING WATER STANDARD**

Chemical or Constituent (and reporting units)	Sample Date	Level Detected	Range of Detections	MCL [MRDL]	PHG (MCLG) [MRDLG]	Typical Source of Contaminant
Aluminum (ppm)	3/1/2014	ND		1		Erosion of natural deposits; residue from some surface water treatment processes.
Fluoride (ppm)	3/1/2014	0.13		2		Erosion of natural deposits; water additive, which promotes strong teeth; discharge from fertilizer and aluminum factories.

**TABLE 5 – DETECTION OF CONTAMINANTS WITH A SECONDARY DRINKING WATER STANDARD**

Chemical or Constituent (and reporting units)	Sample Date	Level Detected	Range of Detections	MCL	PHG (MCLG)	Typical Source of Contaminant
Total Dissolved Solids (ppm)		320		1000		Runoff/leaching from natural deposits.
Specific Conductance (micro gms)		530		1600		Substances that form ions when in water; seawater influence.

**TABLE 6 – DETECTION OF UNREGULATED CONTAMINANTS**

Chemical or Constituent (and reporting units)	Sample Date	Level Detected	Range of Detections	Notification Level	Health Effects Language
MTBE	04/21/'14	0			Exhaust from autos, road runoff.

\*Any violation of an MCL, MRDL, or TT is asterisked. Additional information regarding the violation is provided later in this report.

### Additional General Information on Drinking Water

Drinking water, including bottled water, may reasonably be expected to contain at least small amounts of some contaminants. The presence of contaminants does not necessarily indicate that the water poses a health risk. More information about contaminants and potential health effects can be obtained by calling the USEPA's Safe Drinking Water Hotline (1-800-426-4791).

Some people may be more vulnerable to contaminants in drinking water than the general population. Immuno-compromised persons such as persons with cancer undergoing chemotherapy, persons who have undergone organ transplants, people with HIV/AIDS or other immune system disorders, some elderly, and infants can be particularly at risk from infections. These people should seek advice about drinking water from their health care providers. USEPA/Centers for Disease Control (CDC) guidelines on appropriate means to lessen the risk of infection by *Cryptosporidium* and other microbial contaminants are available from the Safe Drinking Water Hotline (1-800-426-4791).

Lead-Specific Language for Community Water Systems: If present, elevated levels of lead can cause serious health problems, especially for pregnant women and young children. Lead in drinking water is primarily from materials and components associated with service lines and home plumbing. Chemeketa Park Mutual Water Company is responsible for providing high quality drinking water, but cannot control the variety of materials used in plumbing components. When your water has been sitting for several hours, you can minimize the potential for lead exposure by flushing your tap for 30 seconds to 2 minutes before using water for drinking or cooking. If you are concerned about lead in your water, you may wish to have your water tested. Information on lead in drinking water, testing methods, and steps you can take to minimize exposure is available from the Safe Drinking Water Hotline or at <http://www.epa.gov/safewater/lead>.

**Summary Information for Violation of a MCL, MRDL, AL, TT,  
or Monitoring and Reporting Requirement**

<b>VIOLATIONS OF A MCL, MRDL, AL, TT, OR MONITORING AND REPORTING REQUIREMENT</b>				
<b>Violation</b>	<b>Explanation</b>	<b>Duration</b>	<b>Actions Taken to Correct the Violation</b>	<b>Health Effects Language</b>
<b>None</b>				
<b>None</b>				

**For Water Systems Providing Ground Water as a Source of Drinking Water**

<b>TABLE 7 – SAMPLING RESULTS SHOWING FECAL INDICATOR-POSITIVE GROUND WATER SOURCE SAMPLES</b>					
<b>Microbiological Contaminants (complete if fecal-indicator detected)</b>	<b>Total No. of Detections</b>	<b>Sample Dates</b>	<b>MCL [MRDL]</b>	<b>PHG (MCLG) [MRDLG]</b>	<b>Typical Source of Contaminant</b>
<i>E. coli</i>	None		0	(0)	Human and animal fecal waste
Enterococci	None		TT	n/a	Human and animal fecal waste
Coliphage	None		TT	n/a	Human and animal fecal waste

**Summary Information for Fecal Indicator-Positive Ground Water Source Samples,  
Uncorrected Significant Deficiencies, or Ground Water TT**

<b>SPECIAL NOTICE OF FECAL INDICATOR-POSITIVE GROUND WATER SOURCE SAMPLE</b>				
<b>SPECIAL NOTICE FOR UNCORRECTED SIGNIFICANT DEFICIENCIES</b>				
<b>VIOLATION OF GROUND WATER TT</b>				
<b>TT Violation</b>	<b>Explanation</b>	<b>Duration</b>	<b>Actions Taken to Correct the Violation</b>	<b>Health Effects Language</b>
<b>None</b>				
<b>None</b>				

**For Systems Providing Surface Water as a Source of Drinking Water**

<b>TABLE 8 - SAMPLING RESULTS SHOWING TREATMENT OF SURFACE WATER SOURCES</b>	
Treatment Technique <sup>(a)</sup> (Type of approved filtration technology used)	
Turbidity Performance Standards <sup>(b)</sup> (that must be met through the water treatment process)	Turbidity of the filtered water must: 1 – Be less than or equal to <u>0.1</u> NTU in 95% of measurements in a month. 2 – Not exceed <u>0.05</u> NTU for more than eight consecutive hours. 3 – Not exceed <u>1.00</u> NTU at any time.
Lowest monthly percentage of samples that met Turbidity Performance Standard No. 1.	100%
Highest single turbidity measurement during the year	.190
Number of violations of any surface water treatment requirements	None

(a) A required process intended to reduce the level of a contaminant in drinking water.

(b) Turbidity (measured in NTU) is a measurement of the cloudiness of water and is a good indicator of water quality and filtration performance. Turbidity results which meet performance standards are considered to be in compliance with filtration requirements.

\* Any violation of a TT is marked with an asterisk. Additional information regarding the violation is provided below.

**Summary Information for Violation of a Surface Water TT**

<b>VIOLATION OF A SURFACE WATER TT</b>				
<b>TT Violation</b>	<b>Explanation</b>	<b>Duration</b>	<b>Actions Taken to Correct the Violation</b>	<b>Health Effects Language</b>

## Drinking Water Contaminants Detected During Most Recent Samplings

### Regulated Contaminants with Primary Standards—Mandatory Health-Related Standards<sup>1</sup>

Parameter	Units	Secondary PHG or		Storage Tanks	Moody Gulch	Los Gatos Creek	Typical Sources
		MCL	MCLG	(treated) 2/28/2014	{untreated} 2/28/2014	{untreated} 2/28/2014	
<b>Radioactive Radium 228 (Sampled 6/14)</b>	PCi/L	2	N/A	ND	ND	ND	
<b>Inorganic Chemicals</b>							
Aluminum	ppm	1000	600	99	ND	ND	Erosion of natural deposits, residue from some surface water treatment processes
Antimony	ppm	0.006	None	ND	ND	ND	Refineries: fire retardants; ceramics; electronics; solder
Arsenic	ppm	0.01	None	ND	ND	ND	Erosion of natural deposits
Barium	ppm	1	0.02	ND	ND	ND	Erosion of natural deposits
Beryllium	ppm	0.004	0.001	ND	ND	ND	Discharge from metal refineries and from electrical, aerospace and defense industries
Cadmium	ppm	0.005	0.003	ND	ND	ND	Internal corrosion of galvanized pipes; erosion of natural deposits; runoff from waste batteries and paints
Chromium	ppm	0.05	0.01	ND	ND	ND	Erosion of natural deposits.
Copper	ppm	1	0.17	ND	ND	ND	Erosion of natural deposits. Leaching from wood preservatives.
Cyanide	ppm	0.2	0.15	ND	ND	ND	Discharge from steel/metal, plastic and fertilizer factories
Fluoride	ppm	1	1	0.27	0.29	0.31	Erosion of natural deposits
Mercury	ppm	0.002	0.001	ND	ND	ND	Erosion of natural deposits; discharge from refineries and factories; runoff from land fills and cropland
Nickel	ppm	0.1	0.012	ND	ND	ND	Erosion of natural deposits; discharge from metal
Nitrate (NO <sub>3</sub> )	ppm	1.6	45	1.3	1.3	6.8	Runoff and leaching from fertilizer use, erosion of natural deposits
Selenium	ppm	0.05	0.05	ND	ND	ND	Erosion of natural deposits
Thallium	ppm	0.002	0.001	ND	ND	ND	Discharge from electronics, glass and drug factories

Parameter	Units	Secondary PHG or		Storage Tanks	Moody Gulch	Los Gatos Creek	Typical Sources
		MCL	MCLG	(treated) 2/28/2014	{untreated} 2/28/2014	{untreated} 2/28/2014	
Color	-	15 units	None	ND	13.0	ND	Naturally occurring organic materials.
Hardness (CaCO <sub>3</sub> )	ppm	N/A	None	230	230	320	Erosion of natural deposits
Foaming Agents (MBAS)	ppm	.5	None	N/A	N/A	N/A	Municipal and industrial waste discharges.
Iron	ppm	.3	None	ND	56	ND	Erosion of natural deposits
Manganese	ppm	.05	None	ND	ND	73	Leaching from natural deposits
Odor		3 units	None	ND	2.0	ND	Naturally occurring organic materials.
Silver	ppm	.1 ug/L	None	ND	ND	ND	Industrial discharges
Sodium	ppm	N/A	None	30	29	29	Erosion of natural deposits
Turbidity		5 units		.42	0.56	9.4	
Zinc	ppm	5	None	ND	ND	0.1	Runoff and erosion of natural deposits
Total Dissolved Solids	ppm	1000	None	330	340	450	Runoff and erosion of natural deposits
Chloride	ppm	250	None	15	15	20	Erosion of natural deposits
Sulfate	ppm	250	None	84	110	120	Erosion of natural deposits
Boron	ppm	N/A	N/A	ND	110	110	

### Summary Information for Operating Under a Variance or Exemption

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